



Production of sustainable fuels and chemicals from waste gas streams

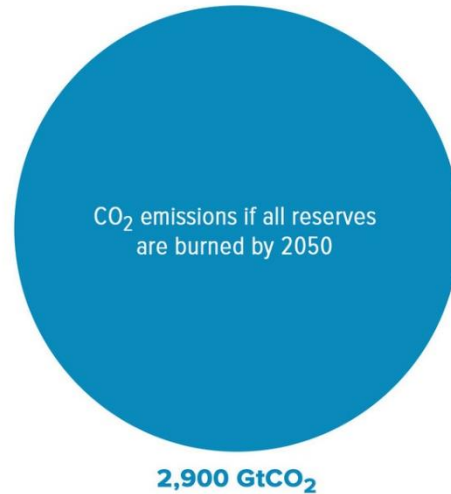
Rasmus Jensen
Senior Scientist

Symposium on Bio-Fuels and Chemicals
La Jolla, CA
30th April 2015

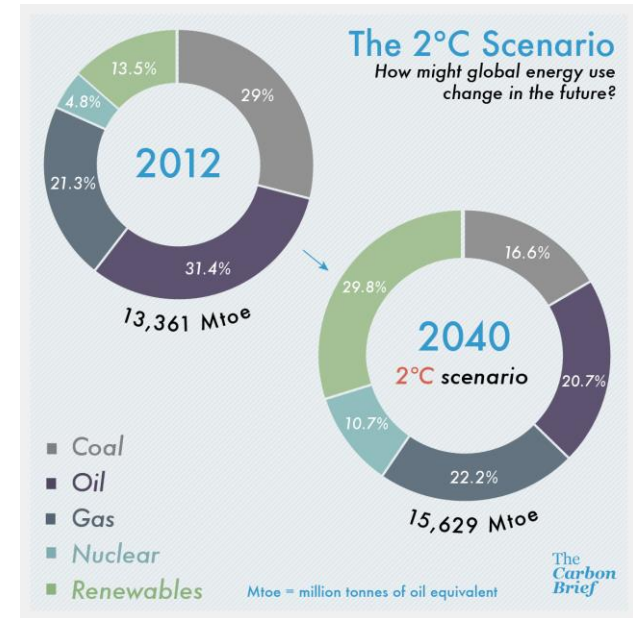
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Staying within 2 degrees



Source: Christophe McGlade & Paul Ekins; U.N. IPCC



80% Coal Reserves
Untouched until 2050

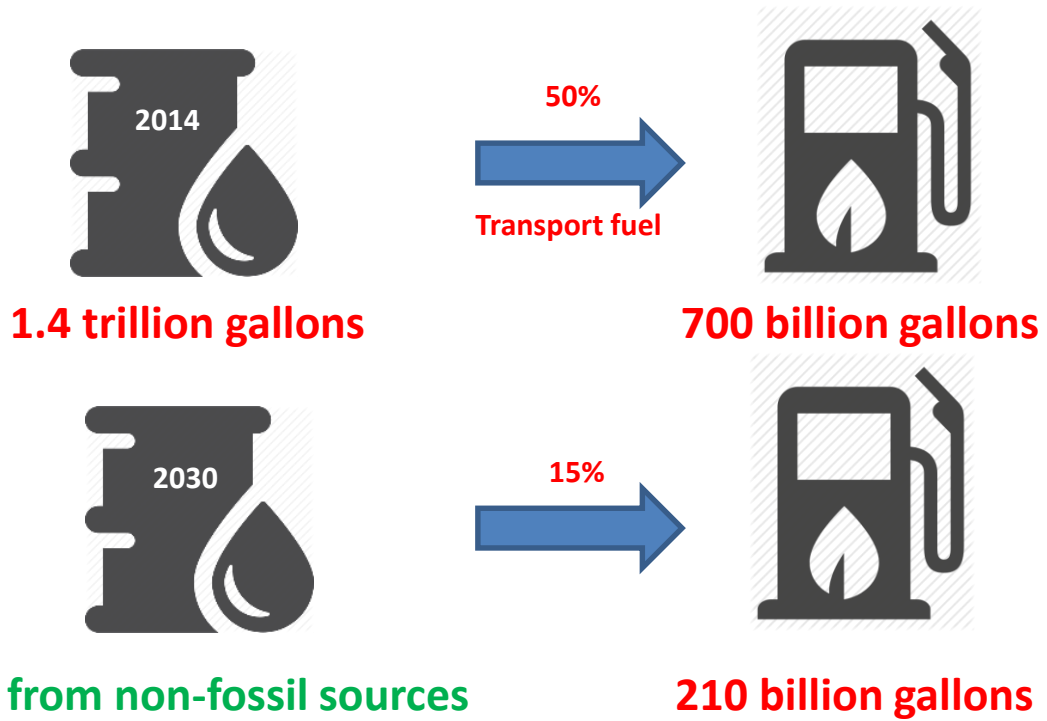
50% Gas Reserves
Untouched until 2050

33% Oil Reserves
Untouched until 2050

Sources:
Nature Geoscience 7, 709–715 (2014)
U.N. IPCC Climate Change 2014 Synthesis Report: <http://www.ipcc.ch/report/ar5/syr/>



What does this mean for fuels?



Only 14% of the 2 degree target

Biofuels today
Additional 1000
<30 B...
200 M gpy facilities
by 2030
cellulosic
gallons

Source: EIA short term energy outlook February 2015

Source: UNEP, 2011 Bridging the Emissions Gap

Based on 100M/G capacity plants

The remaining 86%

**TODAYS TECHNOLOGIES ARE NOT ENOUGH
NEW FEEDSTOCKS, NEW APPROACHES ARE NEEDED**



SPEED TO SCALE AND MARKET ARE KEY

How we treat carbon will define our generation



Recycle to Reduce Pressure on Reserves

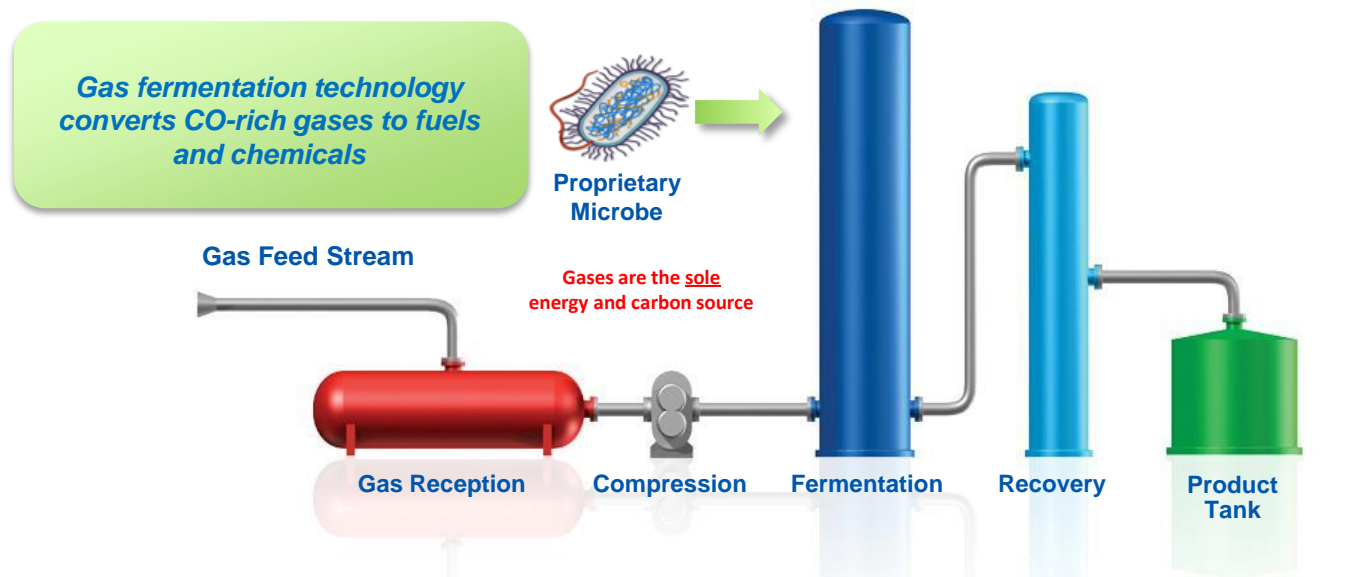


Carbon Reduction through Re-use and Recycling

A 2-degree carbon budget will require countries to leave 80 % of coal, 50 % of gas and 33 % of global oil untouched.

Organization for Economic Growth and Development, Nature

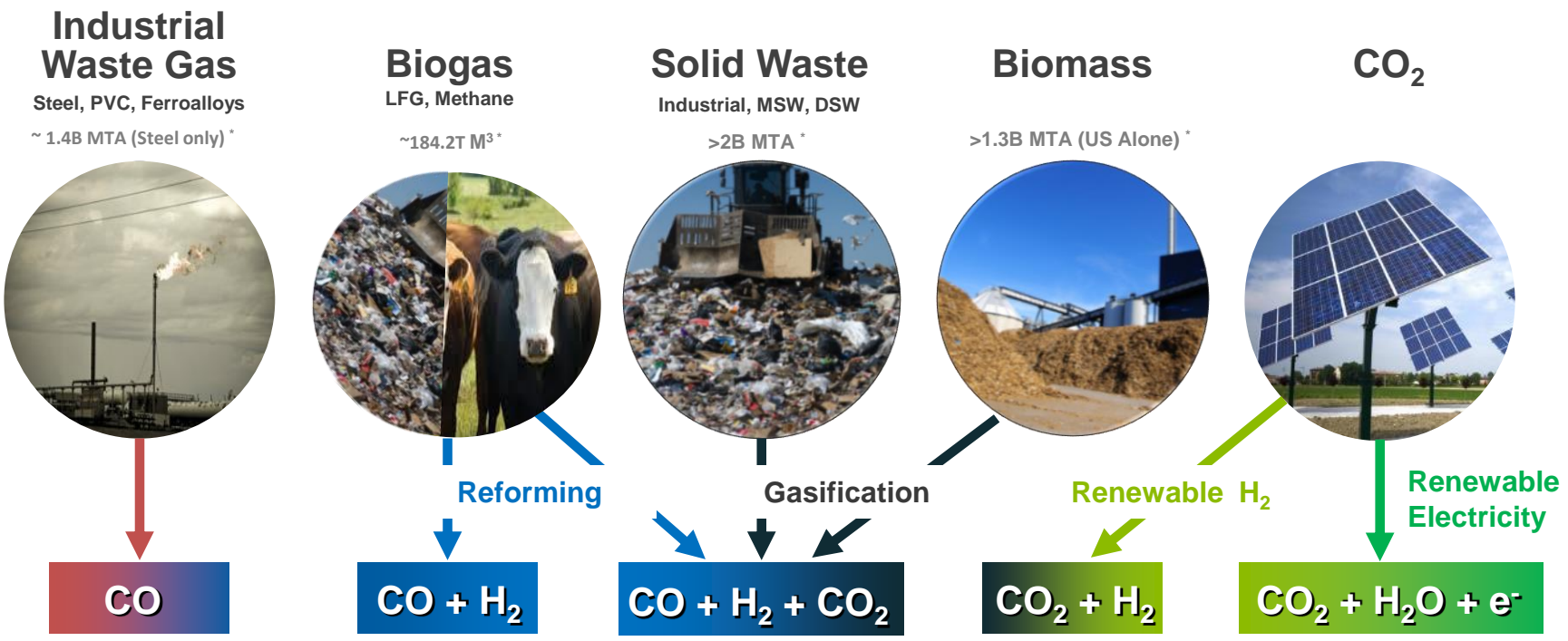
The LanzaTech Process



- Process **recycles** waste carbon into fuels and chemicals
- Process brings underutilized carbon into the fuel pool via **industrial symbiosis**
- Potential to make **material** impact on the future energy pool (>100s of billions of gallons per year)



Waste carbon streams as a Resource



Gas Fermentation

- ✓ Available
- ✓ High Volume/Low Intrinsic Value
- ✓ Most Point Sourced
- ✓ Non-Food

*2010 global production; 2012 proven gas reserves data (IEA, UNEP, IndexMundi, US DOE Billion Ton Update)

Ancient Biology for a Modern Need



The earth is formed

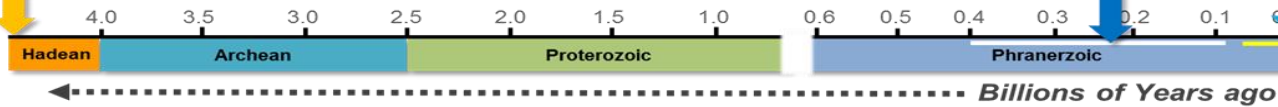


Jurassic Park!



Today

We arrive



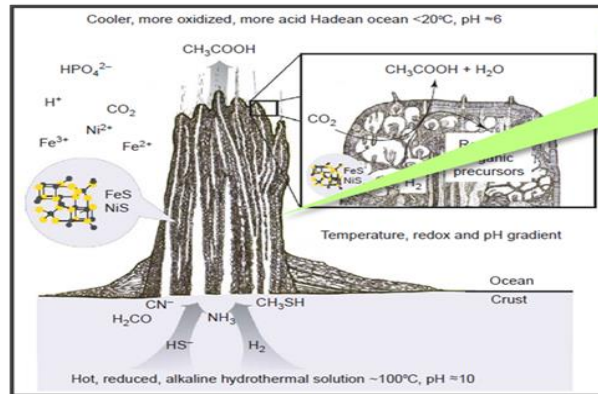
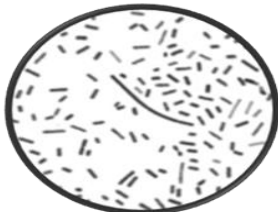
1. Reduced atmosphere

2. CO₂-rich atmosphere

3. O₂-rich atmosphere



Life begins on earth!
Gas fermentation



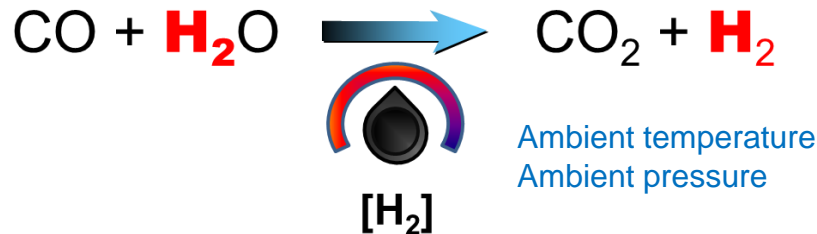
Gases were the only carbon and energy source used by the first life forms.



Unique Chemistry

- **Biological Water-Gas-Shift (WGS) reaction – Making Hydrogen on Demand**

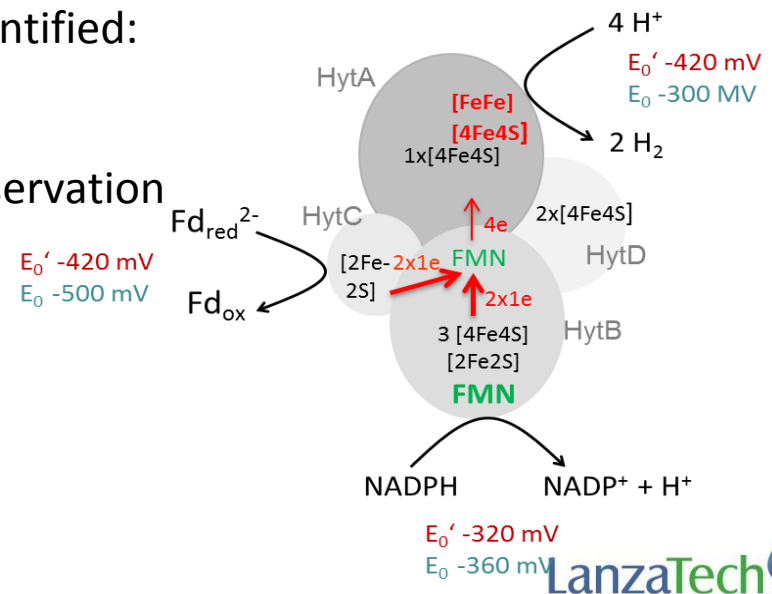
- Carbon Monoxide dehydrogenase (CODH) enzyme:



- Allows any CO:H₂ concentration to be used

- **Electron-Bifurcating enzymes – Coupling of an Uphill with a Downhill Reaction**

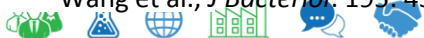
- Two novel Electron-Bifurcating enzymes identified:
 - Nfn Complex and Hyt hydrogenase
- Allows organism to couple with energy conservation
- Allows efficient and reversible reduction of CO₂ with H₂ to formate



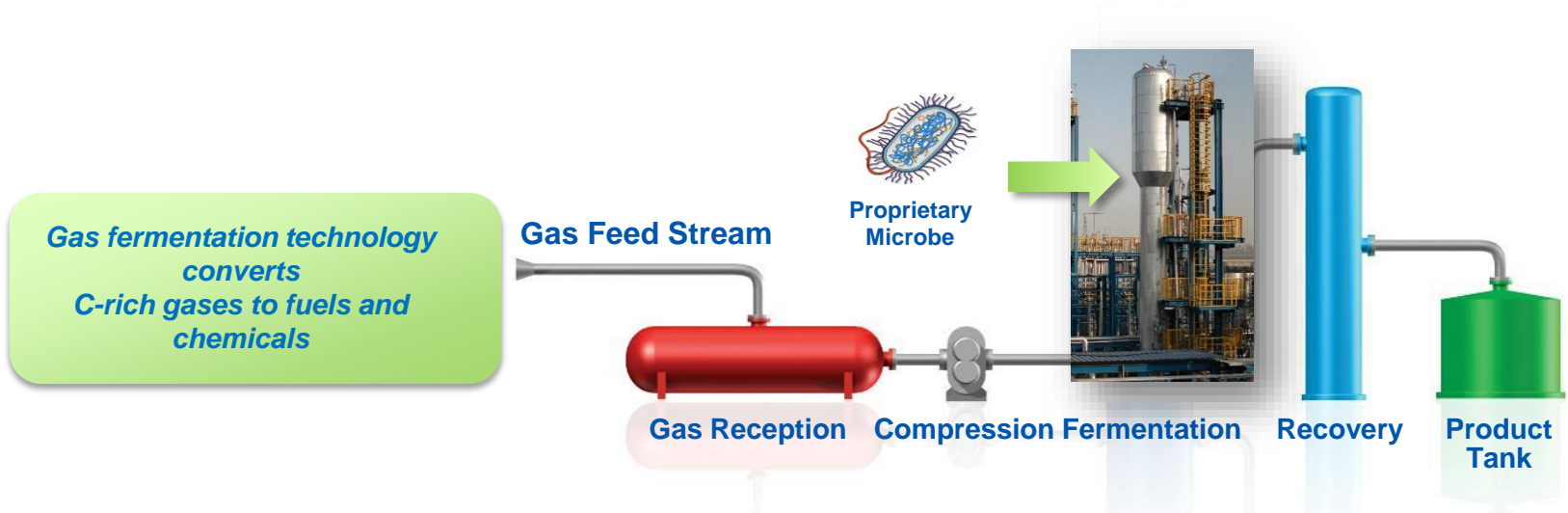
Collaboration with
Thauer lab
@ MPI Marburg

Max-Planck-Institut

Wang et al., *J Bacteriol*: 195: 4373-86



The LanzaTech Process: Ready for Deployment Today



40,000 combined hours on stream
Multiple runs exceeding 2000 hours



Successful Technology Demonstration



- Pre-commercial facility in operation in Shanghai for >8 months meeting and exceeding all its performance targets and milestones
- Capacity 400,000 litres/year ethanol
- Technology has been approved in China for commercial deployment, by the NDRC



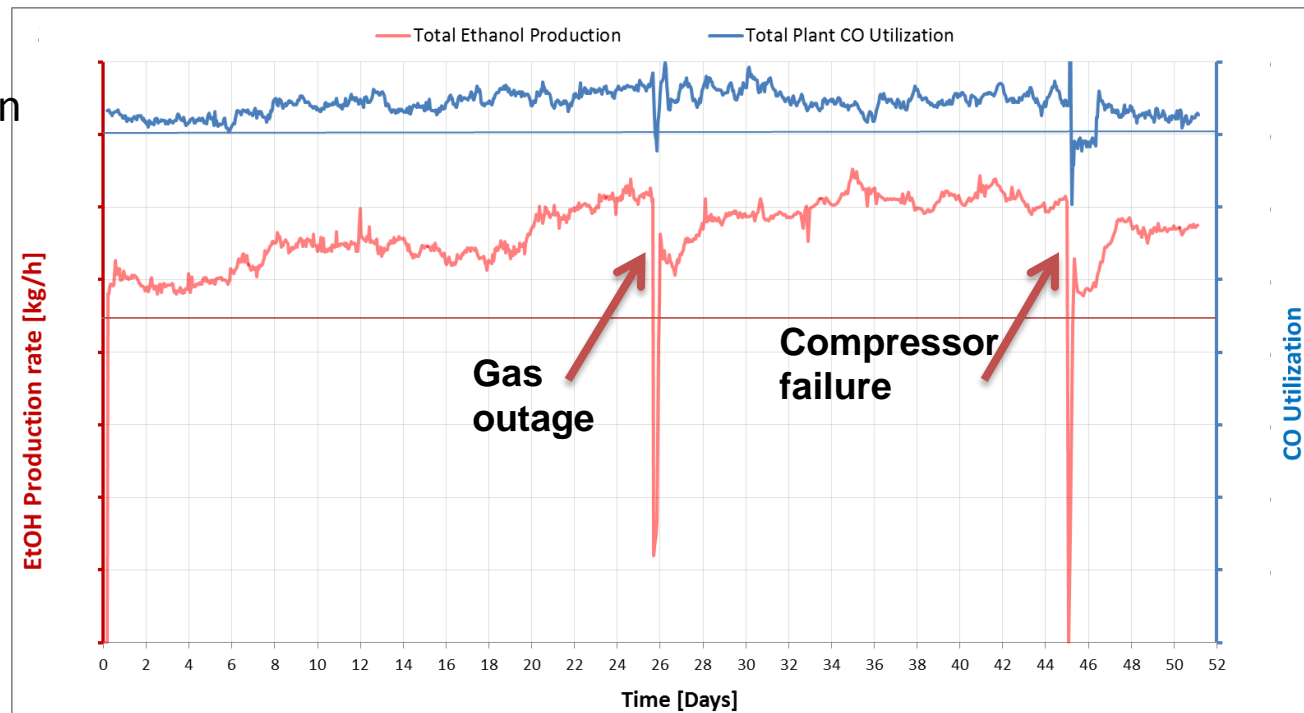
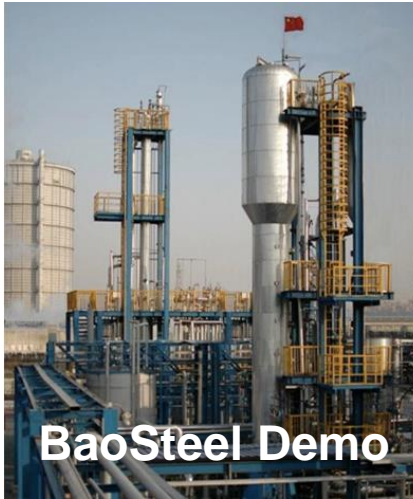
Co-locating LanzaTech's Technology Steel Mill brownfield sites reduces land footprint, improves economics and reduces construction time



- Operation of additional 400,000 litres/year plant with second Chinese Partner, Shougang Group, in Beijing
- Sustainability Assessment of Beijing plant currently underway with RSB.

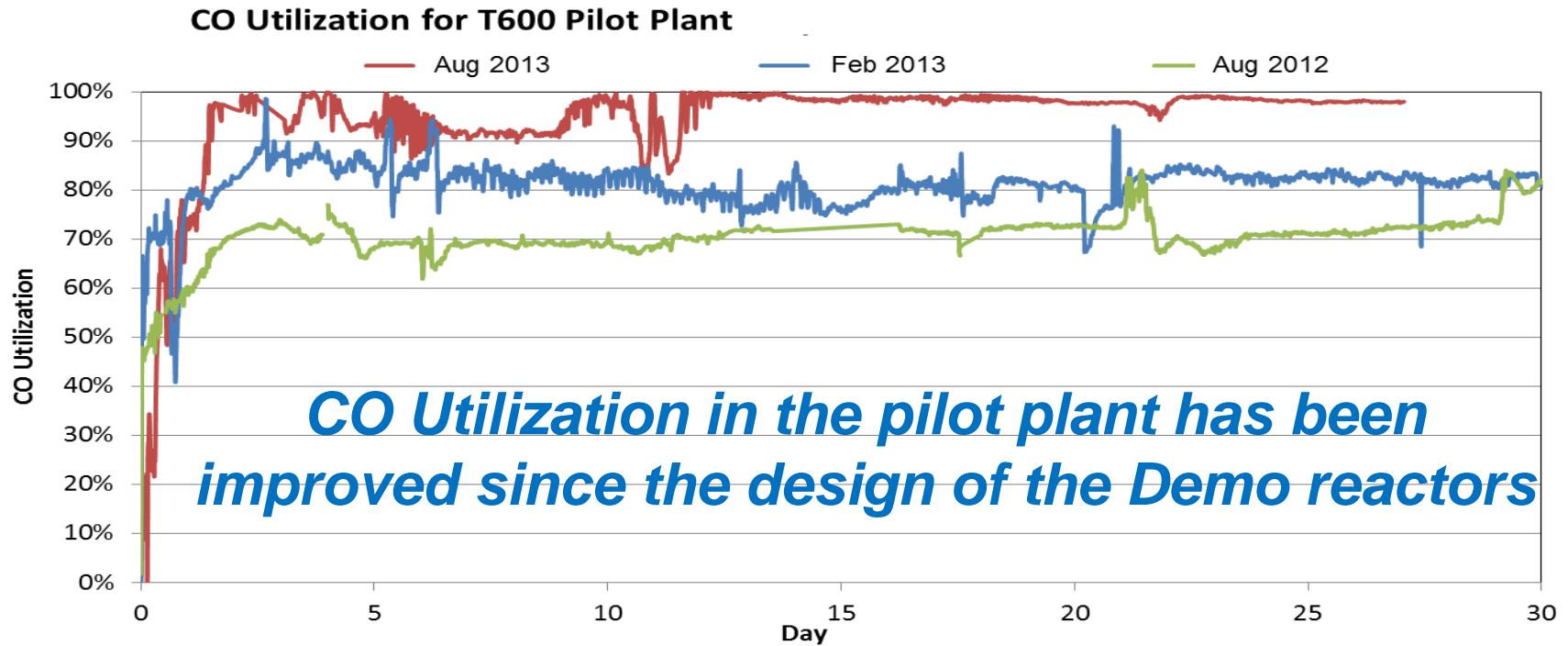
Successful Technology Demonstration

- Successful demonstrations at six industrial sites to date: New Zealand, Asia, United States
- Technology proven using industrial gas, chemicals, utilities, and water
- Over 40,000 total hours on stream
- Continuous runs > 2000 hours



Improvement in Reactor Design & Operation

- CO utilization has improved with advances in reactor design
- LanzaTech Pilot plant operations have shown > 95% CO utilization



MSW to fuel



Pilot plant at gasification site

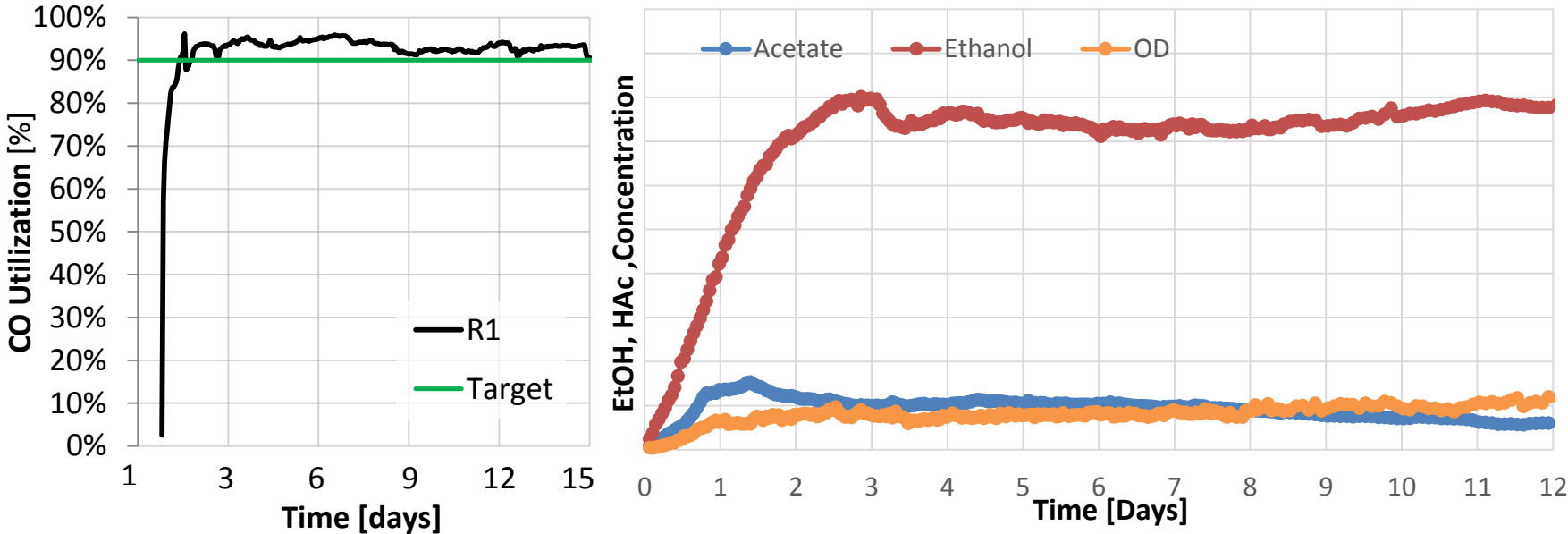


Project overview

LanzaTech has a two year partnership with a major Asian chemical company to convert live-feeds of syngas produced from municipal solid waste (MSW) into ethanol.

LanzaTech has designed, installed, and operates a pilot plant producing ethanol at a MSW processing facility.

Continuous stable ethanol production from MSW

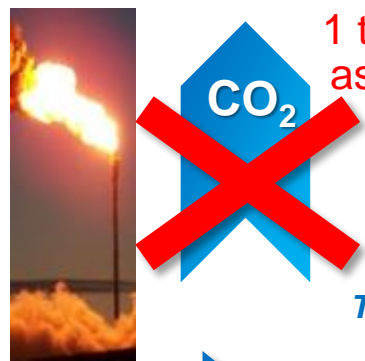


- Continuous with live feeds of MSW Syngas proven
- Operation at commercial ethanol production rates and yields
- Gas utilization efficiency exceeds 90%
- All gas contaminant and variability issues understood and overcome.

LanzaTech are the only company to demonstrate continuous fuel production from MSW syngas.



Why does it matter?

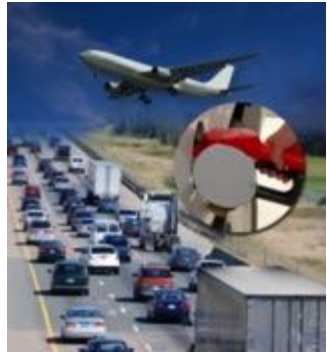


1 tonne ethanol produced as CO averted from flare



5.2 barrels of gasoline are displaced by every tonne of ethanol produced

The LanzaTech Process

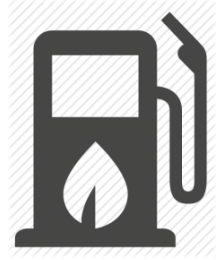


Per tonne of LanzaTech Ethanol

	CO ₂ MT	kg PM	kg NO _x
Averted from flare	2.1	0.6	4.1
Displaced gasoline	+0.5	+2.5	+7.4
Energy required for LanzaTech Process	-0.8	-0.2	-0.8
Avoided per tonne of ethanol	1.8	2.9	10.7

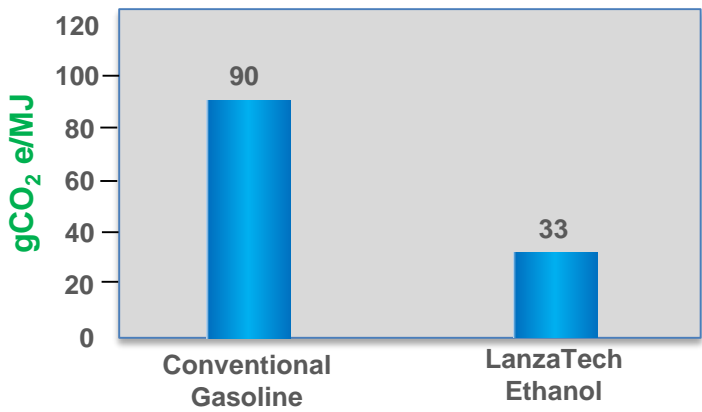
Recycling Waste Gases Produces Low Carbon Fuels

Reduce GHG Emissions

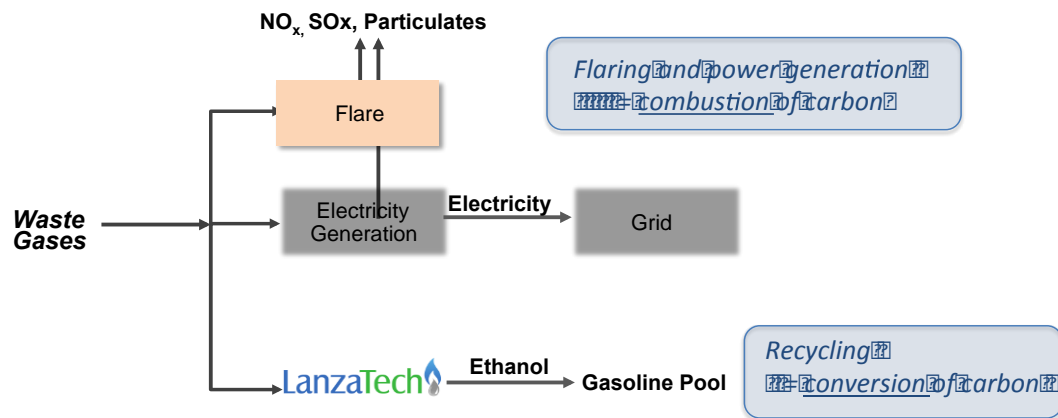


50-70% GHG Reduction over Petroleum Gasoline

Life Cycle GHG Emission



Life Cycle Analyses (LCA) performed in cooperation with , Michigan Tech University,, Roundtable on Sustainable Biomaterials (RSB), E4Tech, and Tsinghua University



Reduce Air Pollutants



>85% reduction in NOx and Particulate Matter compared to combustion at a typical US steel mill

Recycling Waste Gases Impacts GHG and Air Pollution

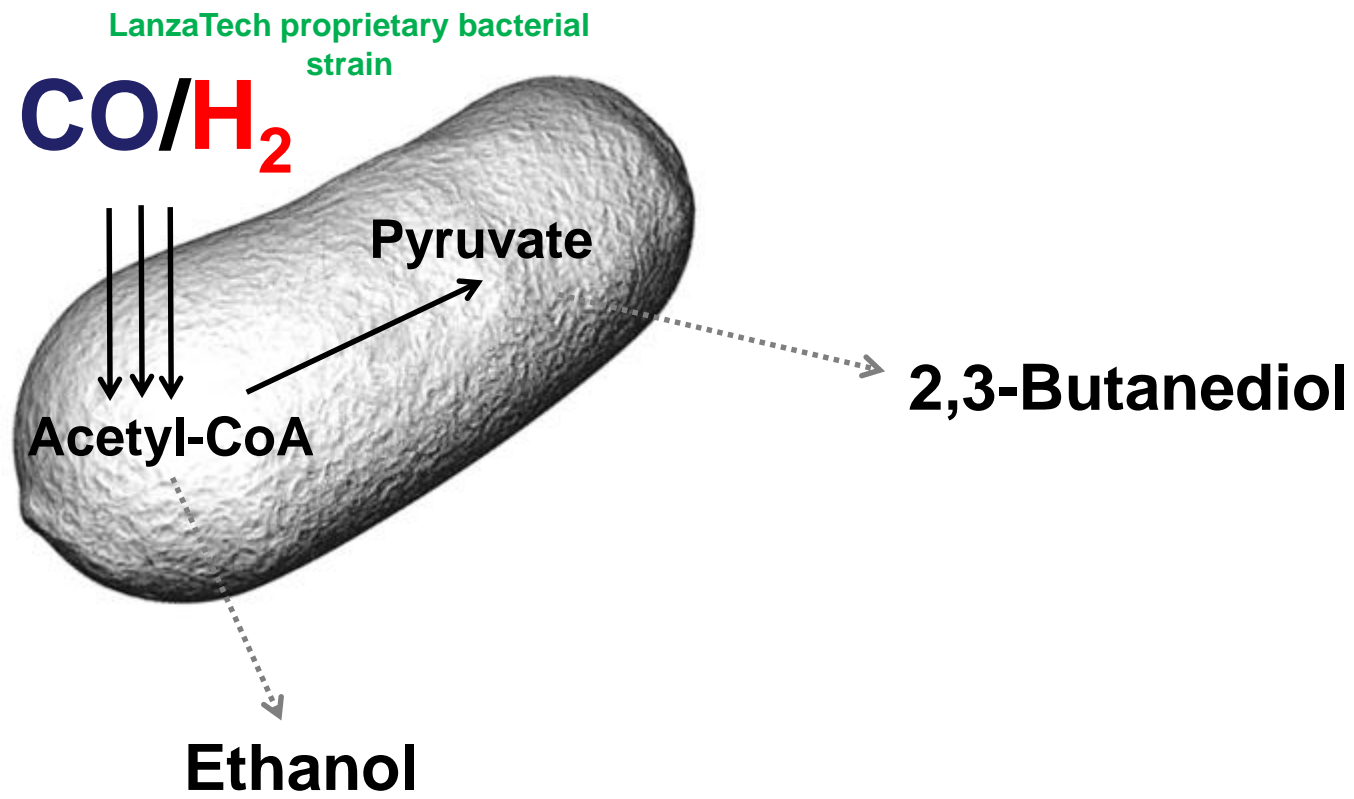
Total Annual Emissions Reduction Potential (China)

	Steel (BF)	Steel (BOF)	Calcium Carbide	FerroAlloy	Phosphorous
CO₂ (M tonnes)	50,500,000	1,700,000	1,800,000	600,000	600,000
PM (tonnes)	70,300	2,400	6,100	800	800
NOx (tonnes)	225,800	7,700	3,700	2,600	2,500
Equivalent Cars	10,600,000	350,000	375,000	125,000	125,000

Utilizing LanzaTech Technology is Equivalent to Removing 11,600,000 Cars from the Road Each Year!



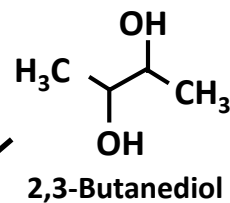
Fuel and Chemicals from Gas



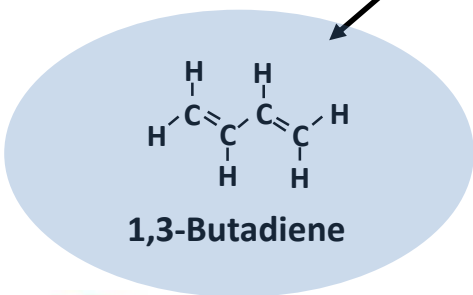
*The LanzaTech gas fermenting microbe can make **both** ethanol and 2,3-butanediol*

C₄ Chemicals from Gases: BDO/Butadiene

Two Step Route:
1. Butanediol production



Direct route:
Developing a Butadiene producing organism

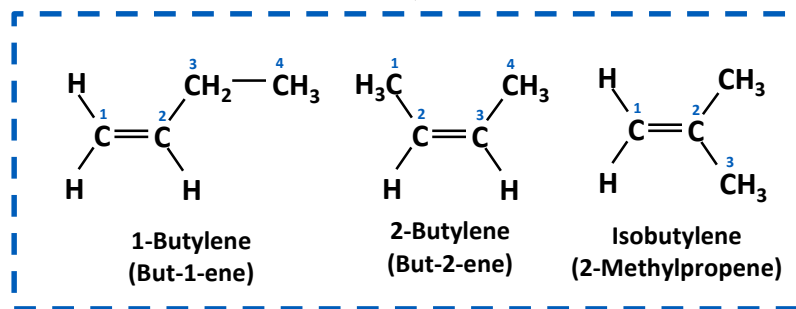


Catalytic Dehydration

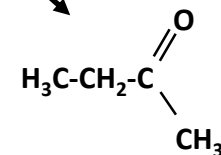
Reductive Elimination

2. Catalytic conversion

Catalytic Dehydration



Butenes

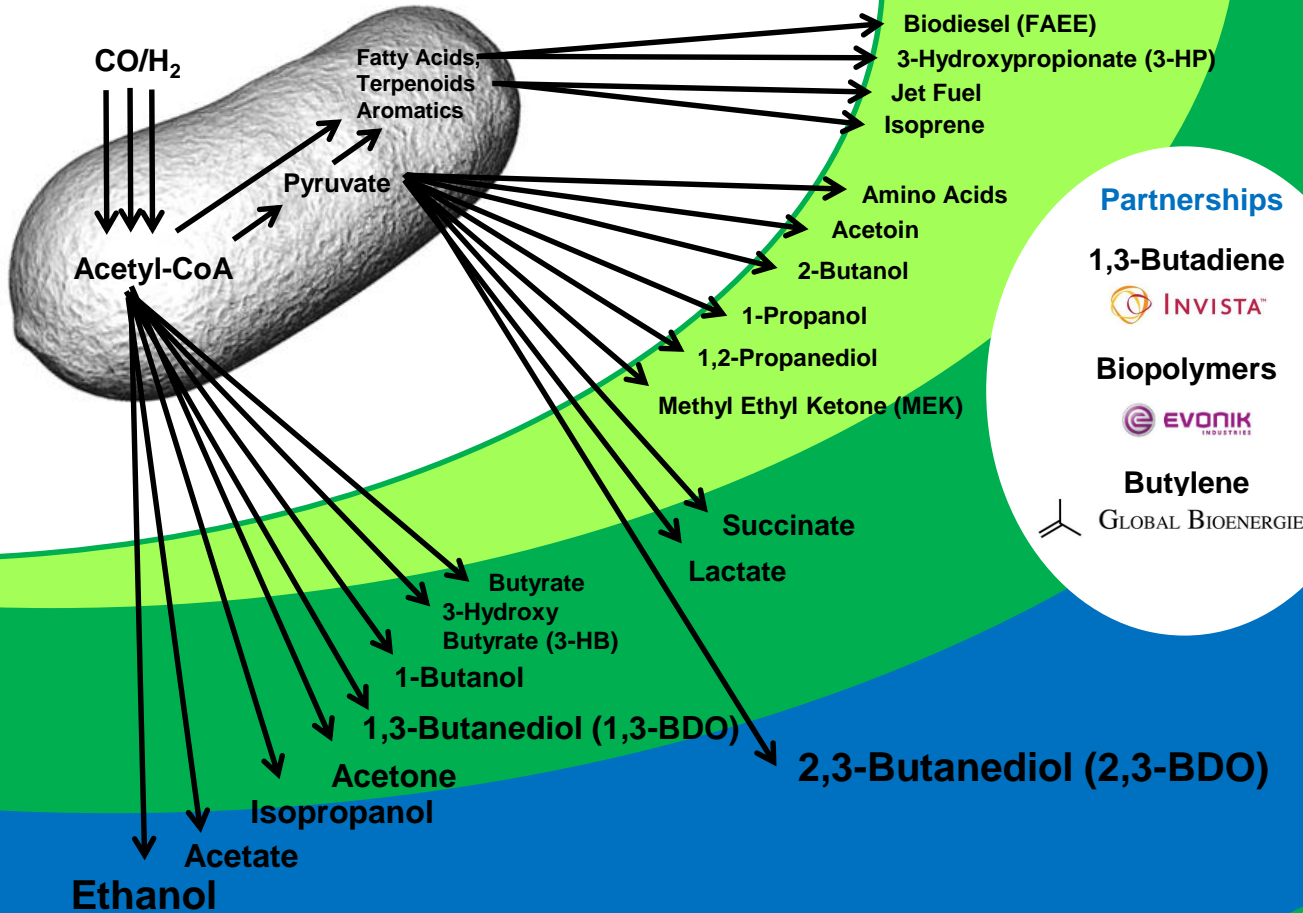


Methyl Ethyl Ketone
(MEK/Butanone)

New Route to C₄s Without Current Supply Challenges



1 Organism, over 25 Products...



Discovery

Lab Scale Process

Scaled-Up Process

Partnerships

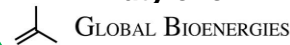
1,3-Butadiene



Biopolymers

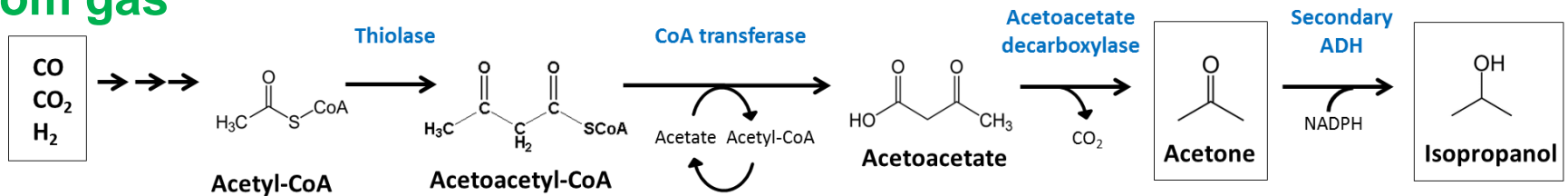


Butylene



Example: Acetone and Isopropanol

1) Demonstrated selective production of either acetone and isopropanol from gas


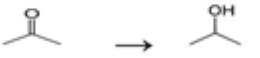



- Integration of ABE fermentation enzymes responsible for conversion of acetyl-CoA to acetone (thiolase, CoA transferase, acetoacetate decarboxylase) into gas fermenting *C. autoethanogenum*
- Identified a novel native primary:secondary alcohol dehydrogenase in *C. auto* for reduction of acetone to isopropanol
 - Highly efficient conversion of acetone to isopropanol
 - KO of enzyme to stop production at acetone

C. autoethanogenum Primary:secondary ADH

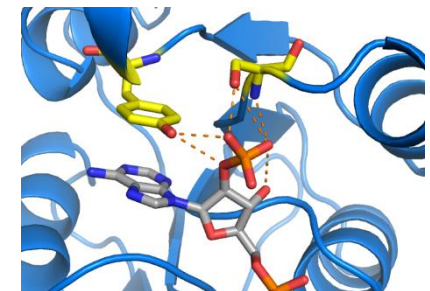
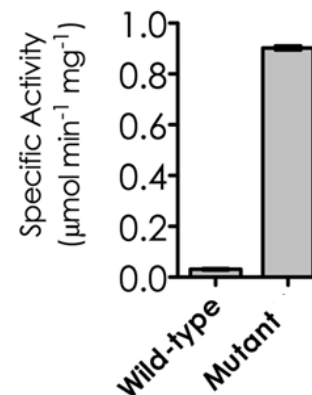
Characterized native primary:secondary ADH enzyme of *C. autoethanogenum*

- Strictly NADPH dependent
- Plays role in conversion of acetaldehyde to ethanol and acetoin to 2,3-butanediol in metabolism of *C. auto*, but highest activity of non-native substrate acetone

	K_{cat} [sec ⁻¹]	K_m [mM]	K_{cat}/K_m [sec ⁻¹ mM ⁻¹]
Primary Adh Acetaldehyde to Ethanol 	93 ± 6	5.5 ± 1.4	1.7 x 10 ⁴
Secondary Adh Acetone to Isopropanol 	51.4 ± 0.8	0.60 ± 0.02	8.6 x 10 ⁴
Acetoin to 2,3-Butanediol 	145 ± 4	71 ± 5	2.0 x 10 ³

Directed evolution to modify selectivity and co-factor requirement

- Switched co-factor from NADPH to NADH
- Changed selectivity



Köpke et al., 2014, AEM 80: 3394-403



In collaboration with
Wayne Patrick lab
@ University of Otago

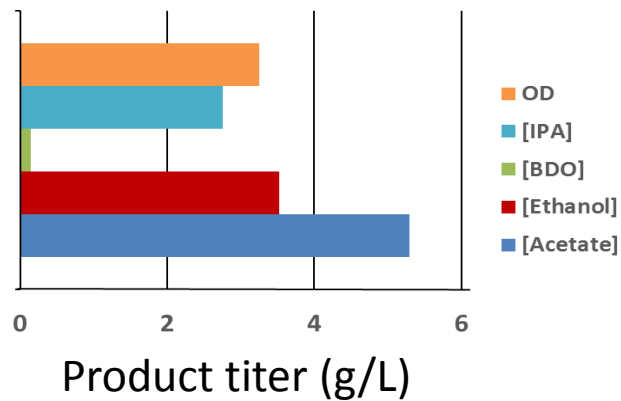


Example: Acetone and Isopropanol

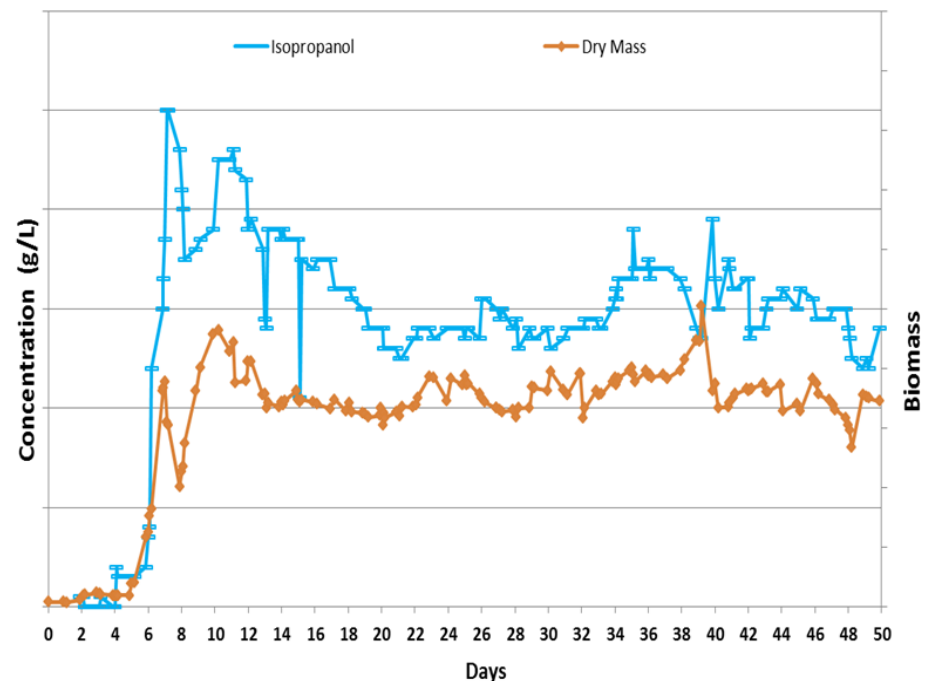
2) Optimizing acetone and isopropanol production in continuous fermentation and elimination of ethanol as byproduct

- Small scale assay developed to screen pathway variants

- Continuous fermentation for 50+ days established and optimized process conditions to improve titer and eliminated ethanol as byproduct



(~44% Isopropanol selectivity)



LanzaTech Global Partnerships



Global Recognition

2015



WBM BIO BUSINESS
AWARD WINNER 2015

*Feedstock of the Year
Award and CEO Jennifer
Holmgren won Business
Person of the Year*



An initiative of the Forum of Young Global Leaders
Sponsored by and in collaboration with: Accenture
Sponsored by: BFI | E.ON | SAP | Shell
Media Partner: Fortinet

*Finalist in the
Entrepreneurship Category of
the Circular Economy
Awards.*

50
Hottest
Companies
in Biofuels
2014-15

*#1 Hottest
Company in
Biofuels and #5 in
Biochemicals*

2014



*Global Cleantech
100 North American
Company of the Year
Award*



*The Guardian
Sustainable Business
Awards 2014
Winner*



*Technical Development
Award from World
Petroleum Council*



*Breakthrough
Innovation Award at
the Platts Global
Metals Award.*

2013



*One of 23 companies
globally with promise
of "significantly
impacting the way
business
and society operate."*



*Listed in Sustainia guide to
innovative solutions
at the forefront of sustainable
transformation.*



*Global Cleantech 100
Continued Excellence
winner.*



*Sustainable Innovation
Award at Platts Global
Energy Awards*



*LanzaTech and Virgin Atlantic named 2013
Observer Ethical Award Winners*



Ancient Biology for Modern Needs

Aligns:



Allows:

